## LETTERS TO THE EDITOR

SYNTHESIS AND REACTIONS OF DERIVATIVES OF ISOINDOLO[2,1-b]-[2,4]BENZODIAZEPINE

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The isoindolo[2,1-b][2,4]benzodiazepine system has been little studied, although other combinations of isoindole and benzodiazepine fragments have been studied in greater detail [1]. Among them, compounds have been found which possess anticonvulsive, sedative, and tranquilizing actions [2-4].

We have shown that the hydrochloride of 2-(o-cyanobenzyl)-1-imino-3H-isoindole (I), mp 246-248°C (from water), obtained by condensing 3-amino-1H-isoindole and o-chloromethyl-benzonitrile in methanol, when it is treated with two equivalents of sodium ethanolate followed by boiling in absolute ethanol for 5 h, undergoes a peculiar molecular cyclization with the formation of the previously unknown 5-imino-5,13-dihydro-11H-isoindolo[2,1-b][2,4]benzo-diazepine (IIa); yield 68%; mp 159-160°C (from nitromethane).

II a R=H; b R=COCH3; c R=CONHC6H5; d R=CSNHC6H5

IR spectrum of (IIa) (KBr),  $cm^{-1}$ : 1635 and 1680 (C=N); 3300 and 3173 (NH). PMR spectrum (from TMS in DMSO), ppm: s, 4.50 (2 H, CH<sub>2</sub>); s, 5.10 (2 H, CH<sub>2</sub>); s, 6.25 (1 H, NH); m, 7.16-7.55 (8 H, aromatic protons).

The hydrogen atom in the imino group of compound (IIa) possesses a definite mobility. Thus, reaction with ketene formed the N-acetyl derivative (IIb), mp  $218-220^{\circ}\text{C}$  (from propan-2-ol); and reaction with phenyl isocyanate and with phenyl isothiocyanate led, respectively, to the phenylurea (IIc), mp  $255-256^{\circ}\text{C}$  (from methanol), and the phenylthiourea (IId), mp  $168-169^{\circ}\text{C}$  (from benzene).

The structures of the compounds synthesized were deduced on the basis of their IR and PMR spectra. The results of their elementary analysis corresponded to the calculated figures.

## LITERATURE CITED

- 1. F. S. Babichev and V. A. Kovtunenko, The Chemistry of Isoindole [in Russian], Naukova Dumka, Kiev (1983).
- 2. G. E. Hardimann, US Patent No. 3,506,647; Chem. Abstr., 73, 3945 (1970).
- 3. G. E. Hardimann, US Patent No. 3,551,445; Chem. Abstr., 74, 141898 (1971).
- 4. Sandos Ltd., Netherlands Patent No. 6,607,814; Chem. Abstr., 67, 43829 (1967).

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